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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1897 Excerpt: .the power of dissolving many substances. There are very few substances, either gaseous, liquid, or solid, that water cannot dissolve. This solvent power is greatly increased by heat, therefore warm water is more useful for washing purposes than cold water. Its action on the soiled clothes is so effective that the particles of dirt and the impure particles from the body are loosened, and the clothes become clean and fresh. We rarely meet with water in its pure state. On account of its great solvent powers it dissolves foreign matters as it runs through the rocks or soil. Hard Water.--When water is impregnated with mineral salts, such as lime or salts of magnesia, the water becomes unfit for laundry purposes, and we call such water hard water. The degree of the hardness of the water depends on the amount of mineral salts the water contains. Sea water is naturally unfit for washing purposes. Soft Water.--Rain water is purer than river, spring, or well water, but yet is not absolutely pure, for in falling it takes in certain gases, such as carbonic acid gas and ammonia, and it also becomes mixed with particles of dirt, smoke, and organic substances which float about in the air. Rain water, however, is free from mineral salts, and on that account we say it is soft. Soft water is particularly valuable for laundry purposes, especially when it has been collected in the country, and is free from the soot and dust...



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